



# HazTech Systems, Inc.

## SAFETY DATA SHEET

**Revision number:** 2  
**Revision date:** 05/06/2015

### 1. IDENTIFICATION

**Product name:** Sulfuric acid, reagent, acs  
**Product code:** RE2304  
**Synonyms:** Oil of Vitriol, Hydrogen Sulfate, Dihydrogen Sulfate  
**CAS:** 7664-93-9  
**RTECS #** WS5600000  
**CI#:** Not available  
**Recommended use:** Fertilizer compositions. Dyes. Echant. Laboratory reagent.  
**Uses advised against:** No information available

**Company:**

HazTech Systems, Inc.  
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 1-800-543-5487 / 1-209-966-8088  
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**Chemical Emergencies:**

HazTech Systems, Inc. (8:00am - 5:00pm) PST  
 1-800-543-5487  
**Transportation Emergencies:**  
 Chemtrec 24-Hour  
 1-800-424-9300 (U.S.A.)  
 1-703-527-3887 (International)

### 2. HAZARD(S) IDENTIFICATION

**Classification**

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200)

Acute toxicity - Inhalation (Gases)	Category 2
Acute toxicity - Inhalation (Vapors)	Category 2
Skin corrosion/irritation	Category 1
Serious eye damage/eye irritation	Category 1

**Label elements**

**Danger**

**Hazard statements**

Fatal if inhaled  
 Causes severe skin burns and eye damage



**Hazards not otherwise classified (HNOC)**

Not Applicable

**Other hazards**

May be harmful if swallowed  
 Reacts violently with water

**Precautionary Statements - Prevention**

Do not breathe dust/fume/gas/mist/vapors/spray  
 Use only outdoors or in a well-ventilated area  
 Wear respiratory protection  
 Wash face, hands and any exposed skin thoroughly after handling  
 Wear protective gloves/protective clothing/eye protection/face protection

**2. HAZARDS IDENTIFICATION**

**Precautionary Statements - Response**

Specific treatment is urgent (see .? on this label)

Immediately call a POISON CENTER or doctor/physician

Specific treatment (see .? on this label)

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

Immediately call a POISON CENTER or doctor/physician.

IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower

Wash contaminated clothing before reuse.

IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing. Immediately call a POISON CENTER or doctor/physician.

IF SWALLOWED: Rinse mouth. DO NOT induce vomiting

**Precautionary Statements - Storage**

Store in a well-ventilated place. Keep container tightly closed

Store locked up

**Precautionary Statements - Disposal**

Dispose of contents/container to an approved waste disposal plant

**3. COMPOSITION/INFORMATION ON INGREDIENTS**

Components	CAS-No.	Weight %	Trade Secret
Sulfuric Acid 7664-93-9	7664-93-9	100	*

**4. FIRST AID MEASURES**

First aid measures

**General Advice:**

Poison information centres in each State capital city can provide additional assistance for scheduled poisons (13 1126). Ensure that medical personnel are aware of the material(s) involved and take precautions to protect themselves. First aider needs to protect himself.

**Skin Contact:**

Wash off immediately with soap and plenty of water. Continue flushing with plenty of water for at least 15 minutes. Remove all contaminated clothes and shoes. Immediate medical attention is required. Call a physician immediately.

**Eye Contact:**

Flush eye with water for 15 minutes. Immediate medical attention is required. Call a physician immediately.

**Inhalation:**

Move to fresh air. If breathing is difficult, give oxygen. If not breathing, give artificial respiration. **WARNING!** It may be hazardous to the person providing aid to give mouth-to-mouth resuscitation when the inhaled or ingested material is toxic, infectious or corrosive. Do not use mouth-to-mouth resuscitation if victim ingested or inhaled the substance; induce artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device. Immediate medical attention is required. Call a physician immediately.

**Ingestion:**

Do not induce vomiting without medical advice. Never give anything by mouth to an unconscious person. If victim is conscious, give water or milk. Follow with Milk of Magnesia or egg whites beaten with water. Immediate medical attention is required. Call a physician or Poison Control Centre immediately.

**Most important symptoms and effects, both acute and delayed**

**Symptoms**

Severe skin and eye irritation or burns.

**Indication of any immediate medical attention and special treatment needed**

**Notes to Physician:**

Treat symptomatically

**Protection of first-aiders**

First-Aid Providers: Avoid exposure to blood or body fluids. Wear gloves and other necessary protective clothing. Dispose of contaminated clothing and equipment as bio-hazardous waste

**5. FIREFIGHTING MEASURES**

Extinguishing Media

**Suitable Extinguishing Media:**

The product is not flammable. If it is involved in a fire, extinguish the fire using an agent suitable for the type of surrounding fire.

**Unsuitable Extinguishing Media:**

No information available.

**Specific hazards arising from the chemical**

**Hazardous Combustion Products:**

No information available.

**5. FIREFIGHTING MEASURES**

**Specific hazards:**

Contact with metals may evolve flammable hydrogen gas  
Metal acetylides (Monocesium and Monorubidium), and carbides ignite with concentrated sulfuric acid.  
White Phosphorous + boiling Sulfuric acid or its vapor ignites on contact.  
May ignite other combustible materials.  
May cause fire when sulfuric acid is mixed with Cyclopentadiene, cyclopentanone oxime, nitroaryl amines, hexalithium disilicide, phosphorous (III) oxide, and oxidizing agents such as chlorates, halogens, permanganates.  
Mixtures of sulfuric acid and any of the following can explode: p-nitrotoluene, pentasilver trihydroxydiaminophosphate, perchlorates, alcohols with strong hydrogen peroxide, ammonium tetraperoxychromate, mercuric nitrite, potassium chlorate, potassium permanganate with potassium chloride, carbides, nitro compounds, nitrates, carbides, phosphorous, iodides, picratres, fulminats, dienes, alcohols (when heated)  
Nitramide decomposes explosively on contact with concentrated sulfuric acid.  
1,3,5-Trinitrosohexahydro-1,3,5-triazine + sulfuric acid causes explosive decomposition.  
Containers may explode when heated or if contaminated with water  
Reaction with water may generate much heat which will increase the concentration of fumes in the air

**Special Protective Actions for Firefighters**

**Specific Methods:**

Water mist may be used to cool closed containers. Do not use water on material itself. Do not get water inside containers. When material is not involved in a fire, do not use water on material itself. DO NOT use combustible materials such as sawdust.

**Special Protective Equipment for Firefighters:**

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear

**6. ACCIDENTAL RELEASE MEASURES**

**Personal precautions, protective equipment and emergency procedures**

**Personal Precautions:**

Keep people away from and upwind of spill/leak. Ensure adequate ventilation. Do not touch damaged containers or spilled material unless wearing appropriate protective clothing. Use personal protective equipment. Avoid contact with skin, eyes and clothing.

**Environmental precautions**

Prevent further leakage or spillage if safe to do so. Should not be released into the environment. Do not let product enter drains. Do not flush into surface water or sanitary sewer system. Prevent entry into waterways, sewers, basements or confined areas.

**Methods and material for containment and cleaning up**

**Methods for containment**

Stop leak if you can do it without risk.

**Methods for cleaning up**

Neutralize with Sodium carbonate or Sodium bicarbonate. Dilute with water. Absorb spill with inert material (e.g. vermiculite, dry sand or earth), then place in a suitable chemical waste container. Clean contaminated surface thoroughly.

**7. HANDLING AND STORAGE**

**Precautions for safe handling**

**Technical Measures/Precautions:**

Use only in area provided with appropriate exhaust ventilation. Do not allow contact with water. Keep away from incompatible materials.

**Safe Handling Advice:**

Wear personal protective equipment. Avoid contact with skin, eyes and clothing. Do not ingest. Do not breathe vapors or spray mist. Handle in accordance with good industrial hygiene and safety practice.

**7. HANDLING AND STORAGE**

**Conditions for safe storage, including any incompatibilities**

**Technical Measures/Storage Conditions:**

Keep container tightly closed in a dry and well-ventilated place. Store at room temperature in the original container. May corrode metallic surfaces. Do not store in uncoated metallic containers. Store in a segregated and approved area. Do not store near combustible materials. Store away from incompatible materials.

**Incompatible Materials:**

Oxidizing agents. Reducing agents. Organic materials. Combustible materials. Bases. Amines. Metals. Water. Acids.

**8. EXPOSURE CONTROL/PERSONAL PROTECTION**

**Control parameters**

**National occupational exposure limits**

**United States**

Components	OSHA	NIOSH	ACGIH	AIHA WHEEL
Sulfuric Acid - 7664-93-9	1 mg/m <sup>3</sup> TWA	1 mg/m <sup>3</sup> TWA	0.2 mg/m <sup>3</sup> TWA thoracic fraction	None

**Canada**

Components	Alberta	British Columbia	Ontario	Quebec
Sulfuric Acid - 7664-93-9	1 mg/m <sup>3</sup> TWA 3 mg/m <sup>3</sup> STEL	0.2 mg/m <sup>3</sup> TWA	0.2 mg/m <sup>3</sup> TWA thoracic	1 mg/m <sup>3</sup> TWAEV 3 mg/m <sup>3</sup> STEV

**Australia and Mexico**

Components	Australia	Mexico
Sulfuric Acid 7664-93-9	3 mg/m <sup>3</sup> STEL 1 mg/m <sup>3</sup> TWA	1 mg/m <sup>3</sup> TWA

**Appropriate engineering controls**

**Engineering measures to reduce exposure:**

Ensure adequate ventilation. Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors and mist below their respective threshold limit value.

**Individual protection measures, such as personal protective equipment**

**Personal Protective Equipment**

**Eye protection:**

Face-shield.

**Skin and body protection:**

Chemical resistant protective suit. Gloves. boots.

**Respiratory protection:**

Vapor respirator. Be sure to use an approved/certified respirator or equivalent.

**Hygiene measures:**

Avoid contact with skin, eyes and clothing. When using, do not eat, drink or smoke. Wash hands before breaks and immediately after handling the product.

**9. PHYSICAL AND CHEMICAL PROPERTIES**

**Physical state:**

Liquid.

**Appearance:**

Viscous. Oily.

**Color:**

Colorless.

**Odor:**

Odorless, but has a choking odor when hot.

**Taste**

Strong. Acid.

**Formula:**

H2SO4

**Molecular/Formula weight:**

98.08

**Flash point (°C):**

No data available

**Flashpoint (°C/°F):**

No information available.

**Flash Point Tested according to:**

Not available

**Lower Explosion Limit (%):**

No information available

**Upper Explosion Limit (%):**

No information available

**Autoignition Temperature (°C/°F):**

No information available

**pH:**

No information available

**Melting point/range(°C/°F):**

No information available

**Boiling point/range(°C/°F):**

270-337 °C/518/638.6 °F

**Decomposition temperature(°C/°F):**

340 °C/644 °F

**Specific gravity:**

1.823-1.84

**Density (g/cm3):**

No information available

**Bulk density:**

No information available

**Vapor pressure @ 20°C (kPa):**

No information available

**Evaporation rate:**

No information available

**Vapor density:**

3.4

**VOC content (g/L):**

No information available

**Odor threshold (ppm):**

No information available

**Partition coefficient**

(n-octanol/water):

No information available

**Viscosity:**

No information available

**Miscibility:**

No information available

**Solubility:**

Easily soluble in cold water  
Sulfuric acid is soluble in water with liberation of much heat  
Soluble in Ethanol

**10. STABILITY AND REACTIVITY**

**Reactivity**

Reacts violently with water

It reacts with alcohols and amines

Incompatible (can react explosively or dangerously) with the following: ACETIC ACID, ACRYLIC ACID, AMMONIUM HYDROXIDE, CRESOL, CUMENE, DICHLOROETHYL ETHER, ETHYLENE CYANOHYDRIN, ETHYLENEIMINE, NITRIC ACID, 2-NITROPROPANE, PROPYLENE OXIDE, SULFOLANE, VINYLIDENE CHLORIDE, DIET HYLENE GLYCOL MONOMETHYL ETHER, ETHYL ACETATE, ETHYLENE CYANOHYDRIN, ETHYLENE GLYCOL MONOETHYL ETHER ACETATE, GLYOXAL, METHYL ETHYL KETONE, dehydrating agents, organic materials, moisture (water), Acetic anhydride, Acetone, cyanohydrin, Acetone+nitric acid, Acetone + potassium dichromate, Acetonitrile, Acrolein, Acrylonitrile, Acrylonitrile+water, Alcohols + hydrogen peroxide, ally compounds such as Allyl alcohol, and Allyl Chloride, 2-Aminoethanol, Ammonium hydroxide, Ammonium triperchromate, Aniline, Bromate + metals, Bromine pentafluoride, n-Butyraldehyde, Carbides, Cesium acetylene carbide, Chlorates, Cyclopentanone oxime, chlorinates, Chlorates + metals, Chlorine trifluoride, Chlorosulfonic acid, 2-cyano-4-nitrobenzenediazonium hydrogen sulfate, Cuprous nitride, p-chloronitrobenzene, 1,5-Dinitronaphthlene + sulfur, Diisobutylene, p-dimethylaminobenzaldehyde, 1,3-Diazidobenzene, Dimethylbenzylcarbinol + hydrogen peroxide, Epichlorohydrin, Ethyl alcohol + hydrogen peroxide, Ethylene diamine, Ethylene glycol and other glycols, , Ethylenimine, Fulminates, hydrogen peroxide, Hydrochloric acid, Hydrofluoric acid, Iodine heptafluoride, Indane + nitric acid, Iron, Isoprene, Lithium silicide, Mercuric nitride, Mesityl oxide, Mercury nitride, Metals (powdered), Nitromethane, Nitric acid + glycerides, p-Nitrotoluene, Pentasilver trihydroxydiaminophosphate, Perchlorates, Perchloric acid, Permanganates + benzene, 1-Phenyl-2-methylpropyl alcohol + hydrogen peroxide, Phosphorus, Phosphorus isocyanate, Picrates, Potassium tert-butoxide, Potassium chlorate, Potassium Permanganate and other permanganates, halogens, amines, Potassium Permanganate + Potassium chloride, Potassium Permanganate + water, Propiolactone (beta)-, Pyridine, Rubidium acetylene carbide, Silver permanganate, Sodium, Sodium carbonate, sodium hydroxide, Steel, styrene monomer, toluene + nitric acid, Vinyl acetate, Thallium (I) azidodithiocarbonate, Zinc chlorate, Zinc Iodide, azides, carbonates, cyanides, sulfides, sulfites, alkali hydrides, carboxylic acid anhydrides, nitriles, olefinic organics, aqueous acids, cyclopentadiene, cyano-alcohols, metal acetylides,

Evolves flammable hydrogen gas on contact with metals

Concentrated sulfuric acid oxidizes, dehydrates, or sulfonates most organic compounds

**Chemical stability**

**Stability:** Stable at normal conditions

**Possibility of Hazardous Reactions:** Hazardous polymerization does not occur

**Conditions to avoid:** Incompatible materials. Exposure to water.

**Incompatible Materials:** Oxidizing agents. Reducing agents. Organic materials. Combustible materials. Bases. Amines. Metals. Water. Acids.

**Hazardous decomposition products:** Sulphur oxides.

**Other Information**

**Corrosivity:** Extremely corrosive in presence of aluminum  
Extremely corrosive in presence of copper  
Extremely corrosive in the presence of stainless steel (316)  
Highly corrosive in the presence of stainless steel (304)  
Non-corrosive in the presence of glass  
Minor corrosive effect on bronze  
Non-corrosive to lead and mild steel  
No corrosion data on brass or zinc

**Special Remarks on Corrosivity:** No information available

**11. TOXICOLOGICAL INFORMATION**

**Information on likely routes of exposure**

**Principal Routes of Exposure:**

Skin. Inhalation. Ingestion.

**Acute Toxicity**

**Component Information**

*Sulfuric Acid - 7664-93-9*

**LD50/oral/rat** = 2140 mg/kg Oral LD50 Rat

**LD50/oral/mouse** = No information available

**LD50/dermal/rabbit** = No information available

**LD50/dermal/rat** = No information available

**LC50/inhalation/rat** = 347 ppm 1 h

420 ppm 1 h

510 mg/m<sup>3</sup> 2 h

**LC50/inhalation/mouse** = 320 mg/m<sup>3</sup> 2 h

**Other LD50 or LC50 information** = No information available

**11. TOXICOLOGICAL INFORMATION**

**Product Information**

**LD50/oral/rat =**

**VALUE- Acute Tox Oral =** 2140mg/kg

**LD50/oral/mouse =**

**Value - Acute Tox Oral =** No information available

**LD50/dermal/rabbit**

**VALUE-Acute Tox Dermal =** No information available

**LD50/dermal/rat**

**VALUE -Acute Tox Dermal =** No information available

**LC50/inhalation/rat**

**VALUE-Vapor =** No information available

**VALUE-Gas =** 347ppm (1-hr)

**VALUE-Dust/Mist =** 510mg/m<sup>3</sup> (2-hr)

**LC50/Inhalation/mouse**

**VALUE-Vapor =** No information available

**VALUE - Gas =** No information available

**VALUE - Dust/Mist =** 320 mg/m<sup>3</sup> 2 h

**Symptoms**

**Skin Contact:**

Severe skin irritation. Causes skin burns.

**Eye Contact:**

Severe eye irritation. Causes eye burns.

**Inhalation**

Fatal if inhaled. Causes severe irritation of the respiratory tract and mucous membranes with sore throat, coughing, sneezing, shortness of breath, and delayed lung edema. Can cause chemical burns (corrosive action) to the respiratory tract and mucous membranes. Inhalation may be fatal as a result of bronchospasm, inflammation, edema of the larynx and bronchi, chemical pneumonitis, and pulmonary edema. May affect cardiovascular system (hypotension, depressed cardiac output, bradycardia). Circulatory shock/collapse with clammy skin, weak and rapid pulse, shallow respiration, and scanty urine may follow. Ischemic liver and heart lesions, kidney failure may occur several hours after unchecked circulatory collapse. Circulatory shock is often the immediate cause of death. May also affect teeth (changes in teeth and supporting structures - erosion, discoloration).

**Ingestion**

May be harmful if swallowed. Causes digestive or gastrointestinal tract burns. Corrosive to the mouth, throat, and stomach. May cause permanent damage to the digestive tract. May cause perforation of the digestive tract. May cause gastritis. May cause abdominal pain. Ingestion may cause nausea, vomiting. Vomit may resemble "coffee grounds". May cause metabolic acidosis.

**Aspiration hazard**

No information available

**Delayed and immediate effects as well as chronic effects from short and long-term exposure**

**Chronic Toxicity**

Inhalation: Prolonged or repeated inhalation may affect behavior (muscle contraction or spasticity), urinary system (kidney damage), and respiratory system/lungs (pulmonary edema, lung damage/changes in lung function with chronic bronchitis and emphysema), teeth (dental discoloration, erosion).

Skin: Prolonged or repeated skin contact may cause dermatitis.

Eyes: Conjunctivitis is also a common finding with chronic exposure.

**Sensitization:**

No information available

**Mutagenic Effects:**

No information available

**Carcinogenic effects:**

May cause cancer. However, evidence is inconclusive. Cancer Status: The International Agency for Research on Cancer (IARC) has classified "strong inorganic acid mists containing sulfuric acid" as a known human carcinogen, (IARC Group 1). However, this classification applies only to mists containing sulfuric acid generated during an industrial process and not to (almost) pure sulfuric acid or sulfuric acid solutions; The ACGIH has classified "strong inorganic acid mists containing sulfuric acid" as a suspected human carcinogen (ACGIH Group A2). However, this classification applies only to mists containing sulfuric acid generated during an industrial process and not to (almost) pure sulfuric acid or sulfuric acid solutions.

**11. TOXICOLOGICAL INFORMATION**

Components	ACGIH - Carcinogens	IARC	NTP	OSHA HCS - Carcinogens	Australia - Prohibited Carcinogenic Substances	Australia - Notifiable Carcinogenic Substances
Sulfuric Acid	A2 Suspected Human Carcinogen (contained in strong inorganic acid mists)	Group 1 - Monograph 54 [1992] Occupational exposure to mists and vapours from sulfuric acid and other strong inorganic acids	Not listed	Present	Not listed	Not listed

**Reproductive toxicity**

No data is available

**Reproductive Effects:**

No information available

**Developmental Effects:**

No information available

**Teratogenic Effects:**

Developmental effects and Teratogenicity: According to the Registry of Toxic Effects of Chemical Substances (RTECS reference - Murry et al, "Embryotoxicity of Inhaled Sulfuric Acid Aerosol in Mice and Rabbits", Journal of Environmental Science and Health, Part C, Vol. 13, pages 251-266, 1979), musculoskeletal developmental abnormalities were found in rabbits at a dose of 20 mg/m<sup>3</sup> for 7 hrs. However, REPROTOX and Shepard's Catalog of Teratogenic Agents, citing this same study, stated that inhalation of sulfuric acid fumes did not increase congenital anomalies in the offspring of treated pregnant mice or rabbits. Furthermore, the Hazard Substance Data Bank (HSDB) also stated that in a developmental toxicity study conducted under a method similar to OECD test Guideline 414 that no significant effects on mean numbers of implants/dam, live fetuses/litter or resorptions/litter were observed in mice and rabbits exposed by inhalation to sulfuric acid aerosol at 5 and 20 mg/m<sup>3</sup> during gestation and therefore could not be considered embryotoxic, or fetotoxic.

**Specific Target Organ Toxicity**

**STOT - single exposure**

No information available

**STOT - repeated exposure**

No information available

**Target Organs:**

Skin. Eyes. Respiratory system. Teeth.

**12. ECOLOGICAL INFORMATION**

**Ecotoxicity**

**Ecotoxicity effects:**

Aquatic environment.

*Sulfuric Acid - 7664-93-9*

**Freshwater Fish Species Data:**

500 mg/L LC50 Brachydanio rerio 96 h static 1

**Persistence and degradability:**

No information available

**Bioaccumulative potential:**

No information available

**Mobility:**

No information available

**13. DISPOSAL CONSIDERATIONS**

**Disposal Methods**

**Waste from residues / unused products:**

Waste must be disposed of in accordance with Federal, State and Local regulation.

**Contaminated packaging:**

Empty containers should be taken for local recycling, recovery or waste disposal

Components	RCRA - F Series Wastes	RCRA - K Series Wastes	RCRA - P Series Wastes	RCRA - U Series Wastes
Sulfuric Acid	None	None	None	None

**14. TRANSPORT INFORMATION**

<p><b>DOT</b></p> <p>UN-No: UN1830          Proper Shipping Name: Sulfuric acid          Hazard Class: 8          Subsidiary Risk: Not applicable          Packing Group: II          Marine Pollutant: No data available          ERG No: 137          DOT RQ (lbs): No information available          Symbol(s): R4</p> <p><b>TDG (Canada)</b></p> <p>UN-No: UN1830          Proper Shipping Name: Sulfuric acid          Hazard Class: 8          Subsidiary Risk: No information available          Packing Group: II          Description: No information available</p> <p><b>ADR</b></p> <p>UN-No: UN1830          Proper Shipping Name: Sulphuric acid          Hazard Class: 8          Packing Group: II          Subsidiary Risk: No information available          Classification Code: No information available          Description: No information available          CEFIC Tremcard No: No information available</p> <p><b>IATA</b></p> <p>UN-No: UN1830          Proper Shipping Name: Sulphuric acid          Hazard Class: 8          Subsidiary Risk: No information available          Packing Group: II          ERG Code: 8L          Description: No information available</p>	<p><b>IMO / IMDG</b></p> <p>UN-No: UN1830          Proper Shipping Name: Sulphuric acid          Hazard Class: 8          Subsidiary Risk: No information available          Packing Group: II          Description: No information available          IMDG Page: No information available          Marine Pollutant: No information available          EMS: F-A          MFAG: No information available          Maximum Quantity: No information available</p> <p><b>RID</b></p> <p>UN-No: UN1830          Proper Shipping Name: Sulphuric acid          Hazard Class: 8          Subsidiary Risk: 8          Packing Group: II          Classification Code: No information available          Description: No information available</p> <p><b>ICAO</b></p> <p>UN-No: UN1830          Proper Shipping Name: Sulphuric acid          Hazard Class: 8          Subsidiary Risk: No information available          Packing Group: II          Description: No information available</p>
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**15. REGULATORY INFORMATION**

**International Inventories**

Components	U.S. TSCA	KOREA KECL	Philippines (PICCS)	Japan ENCS	CHINA	Australia (AICS)	EINECS-No.
<i>Sulfuric Acid</i>	Present	Present KE-32570	Present	Present (1)-724 (1)-430	Present	Present	Present 231-639-5

**U.S. Regulations**

*Sulfuric Acid*

- Massachusetts RTK: Present
- New Jersey RTK Hazardous Substance List: Present
- New Jersey (EHS) List: Present
- New Jersey - Discharge Prevention - List of Hazardous Substances: Present
- Pennsylvania RTK: Environmental hazard
- Pennsylvania RTK - Environmental Hazard List: Present
- Minnesota - Hazardous Substance List: Present
- New York Release Reporting - List of Hazardous Substances:
  - 1000 lb RQ
  - 100 lb RQ
- Louisiana Reportable Quantity List for Pollutants: 1000lbfinal RQ  
454kgfinal RQ
- California Directors List of Hazardous Substances: Present
- FDA - Food Additives Generally Recognized as Safe (GRAS): 21 CFR 184.1095
- California Prop. 65: Safe Drinking Water and Toxic Enforcement Act of 1986.
- Chemicals Known to the State of California to Cause Cancer:

WARNING: This product contains a chemical known to the State of California to cause cancer. (See table on next page)

**Chemicals Known to the State of California to Cause Reproductive Toxicity:**

This product does not contain a chemical requiring a warning under California Prop. 65. (See table on next page)



**15. REGULATORY INFORMATION**

Components	Carcinogen	Developmental Toxicity	Male Reproductive Toxicity	Female Reproductive Toxicity:
Sulfuric Acid	Listed under strong inorganic mists containing sulfuric acid	Not Listed	Not Listed	Not Listed

**CERCLA/SARA**

Components	CERCLA - Hazardous Substances and their Reportable Quantities	Section 302 Extremely Hazardous Substances and TPQs	Section 302 Extremely Hazardous Substances and RQs	Section 313 - Chemical Category	Section 313 - Reporting <i>de minimis</i>
<i>Sulfuric Acid</i>	1000 lb final RQ 454 kg final RQ	1000 lb TPQ	None	None	1.0 % de minimis concentration

**U.S. TSCA**

Components	TSCA Section 5(a)2 - Chemicals With Significant New Use Rules (SNURS)	TSCA 8(d) -Health and Safety Reporting
<i>Sulfuric Acid</i>	Not Applicable	Not Applicable

**Canada**

**WHMIS hazard class:**

D1A Very toxic materials  
E Corrosive material

**Sulfuric Acid**

D1A E including >51%, <=51%

**Canada Controlled Products Regulation:**

This product has been classified according to the hazard criteria of the CPR (Controlled Products Regulation) and the MSDS contains all of the information required by the CPR.

Components	WHMIS Ingredient Disclosure List -
Sulfuric Acid	1 %

**Inventory**

Components	Canada (DSL)	Canada (NDSL)
Sulfuric Acid	Present	Not Listed

Components	CEPA Schedule I - Toxic Substances	CEPA - 2010 Greenhouse Gases Subject to Mandatory Reporting
Sulfuric Acid	Not listed	Not listed

**EU Classification**

**R-phrases**

R35 - Causes severe burns.

**S-phrases**

S26 - In case of contact with eyes, rinse immediately with plenty of water and seek medical advice.

S30 - Never add water to this product.

S45 - In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).

S 1/2 - Keep locked up and out of the reach of children.

Components	Classification	Concentration Limits:	Safety Phrases
Sulfuric Acid	C; R35	15%<=C: C; R:35 5%<=C<15%: Xi; R:36/38	S1/2 S26 S30 S45

The product is classified in accordance with Annex VI to Directive 67/548/EEC

**Indication of danger:**

C - Corrosive.



16. OTHER INFORMATION

NFPA	HMIS	Personal Protective Equipment
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Health Hazard	3
Fire Hazard	0
Reactivity	2



See Section 8.

Revision Date: 05/06/2015

Prepared by: HazTech Systems, Inc.

*This information is based on HazTech Systems, Inc.'s, current knowledge and is intended to describe the product for the purposes of health, safety and environmental requirements only. It should not therefore be construed as guaranteeing any specific property of the product*